## CH301 Worksheet 1: High School Chemistry Questions.

For each problem, identify the kind of question BEFORE answering it.

- 1. A 25 kg ball has a kinetic energy of 225 J. What is the speed of the ball? *What kind of question is this?*
- 2. How many molecules of water are there in 100 mL of water? (The density of water = 1.000 g/ml.) *What kind of question is this?*
- 3. Given that the atomic mass of carbon is given as 12.0107 g/mol, what are the approximate relative abundances of the two isotopes <sup>12</sup>C and <sup>13</sup>C? *What kind of question is this?*
- 4. In the electrolysis of 2H<sub>2</sub>O → 2H<sub>2</sub> + O<sub>2</sub>, how many atom of O in O<sub>2</sub> are made from 4 g of H<sub>2</sub>O? What kind of question is this?
- 5. What are the most likely ionic forms of (a) sodium and (b) calcium? *What kind of question is this?*
- 6. What is the product of a reaction between metallic lithium and oxygen, O<sub>2</sub>? *What kind of question is this?*
- 7. Name the following species: (a) Cl<sup>-</sup>, (b) NO<sub>3</sub><sup>-</sup>, (c) C<sub>3</sub>H<sub>8</sub>. What kind of question is this?
- 8. The combustion of ethanol has the following equation:  $C_2H_5OH + 3 O_2 \rightarrow 2 CO_2 + 2 H_2O$ What mass of CO<sub>2</sub> is formed from 1 kg of ethanol? *What kind of question is this?*
- 9. Vitamin K contains 78.95% C, 3.95% H, and 21.05% O by mass. What is its empirical formula? *What kind of question is this?*
- 10. Given that vitamin K (see question 9) has molecular weight of 158.15 g/mol, what is its molecular formula? *What kind of question is this?*

- 11. When sand is stirred up in a glass of water a (homogeneous of heterogeneous?) mixture is formed. *What kind of question is this?*
- 12. If 10.0 g of hydrochloric acid is dissolved in enough water to make 1.00 liter of solution, what is the molarity of the solution? *What kind of question is this?*
- 13. Balance the following chemical equation:  $BCl_3 + H_2O \rightarrow B(OH)_3 + HCl_3$ *What kind of question is this?*
- 14. A microwave oven used radiation at a frequency of 2450 MHz to vibrationally excite water molecules in food. What is the wavelength of this radiation? *What kind of question is this?*
- 15. How may protons, neutrons, and electrons are in the most common atomic form of chlorine? *What kind of question is this?*
- 16. If 75 mL of water is added to 25 mL of 1 M H<sub>2</sub>SO<sub>4</sub> in water, what is the new concentration of H<sub>2</sub>SO<sub>4</sub>? *What kind of question is this?*
- 17. Classify the following properties of matter as physical of chemical: (a) color, (b) flammability, (c) density. *What kind of question is this?*

18. Rank the following four forms of electromagnetic radiation in terms of decreasing energy: *Microwave, X-ray, radiofrequency, yellow light. What kind of question is this?* 

- 19. Which of the following is *not* a strong acid: HI, H<sub>2</sub>SO<sub>4</sub>, HCl, or HNO<sub>2</sub>? *What kind of question is this?*
- 20. What is the oxidation number of nitrogen in nitrate? *What kind of question is this?*